# REFLECTION OF CONFORMATIONAL PROPERTIES OF A FREE MOLECULE OF NONMESOGEN 4-PHENYLAZOBENZOIC ACID IN CRYSTAL STRUCTURES 

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## Introduction

The principles of supramolecular chemistry are successfully applied in the creation of new hydrogen bonded mesomorphic complexes with mesogen and nonmesogen complementary compounds. The structure of nonmesogen 4-phenylazobenzoic acid (4-PABA) allows its use as a monofunctional component of potentially mesogen hydrogen bonded complexes.

## Purpose of the work

Quantum-chemical study of the conformational properties of a free 4-PABA molecule, comparison of the obtained data with structures stabilized in the crystal.

| Method |  |  |  |
| :---: | :---: | :---: | :---: |
| Quantum chemical calculations <br> conformational properties) <br> Gaussian09 (DFT/B3LYP/cc-pVTZ) |  |  |  |

## Method

 conformational properties)Gaussian09 (DFT/B3LYP/cc-pVTZ)


Histogram of the distribution of the angle $\mathrm{N}=\mathrm{N} . . . \mathrm{C}=\mathrm{O}$ in the substituted 4-PABA from the CCDC (43 objects)

## Conclusions

Using the DFT method, it was determined that the 4-PABA molecule has 8 conformers, among which the conditional "cis" and "trans " conformers can be distinguished. The relative energy of the "cis" conformers is significantly higher ( $15-22 \mathrm{kcal} / \mathrm{mol}$ ) than the energy of the "trans" conformers. The most energetically advantageous are the "trans" conformers 1 and 2 , which differ from each other in the position of the carboxyl group relative to the plane of the molecule. The torsion angle $\mathrm{N}=\mathrm{N} \ldots \mathrm{C}=\mathrm{O}$ in conformer 1 takes the value $0^{\circ}$, and in conformer $2-180^{\circ}$. This torsion angle is selected as a search criterion in the Cambridge database. It was determined that the crystal contains structures close to conformer 1 (53.5\%) and conformer 2 ( $46.5 \%$ ). No " Cis " conformers were detected. Thus, the calculated data adequately describe the conformational properties of the molecules. Modeling of H-complexes can be based on using the structure of a more favorable conformer.

